Six Types of Reaction Homework

Fill in the blanks with the correct type of reaction:

| 1) | A(n) reaction is one in which oxygen reacts with another compound to form carbon dioxide, water, and heat. |
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| 2) | A(n) reaction takes place when one element trades places with an element in an ionic compound. An example of this reaction is when silver nitrate reacts with copper metal to form copper (I) nitrate and silver metal. |
| 3) | A(n) reaction is one where a complex molecule breaks apart to form two or more less complicated molecules. |
| 4) | A(n) reaction is one where the cations of two ionic compounds switch places. An example of this is when sodium chloride reacts with silver nitrate to form silver chloride and sodium nitrate. |
| 5) | A(n) reaction is when two or more simple compounds join to form a more complicated one. |
| | he following questions, balance the equation and indicate what type of tion the equation is showing: |
| 6) | $_$ LiOH + $_$ H ₂ SO ₄ \rightarrow $_$ Li ₂ SO ₄ + $_$ H ₂ O |
| | Type of reaction: |
| 7) | $C_5H_{12} + C_2 \rightarrow CO_2 + H_2O$ |
| | Type of reaction: |
| 8) | Fe + Ni(OH) ₂ \rightarrow Ni + Fe(OH) ₃ |
| | Type of reaction: |
| 9) | NaHCO ₃ → NaOH + CO ₂ |
| | Type of reaction: |
| 10) | Pb(C ₂ H ₃ O ₂) ₂ + NaBr \rightarrow PbBr ₂ + NaC ₂ H ₃ O ₂ Type of reaction: |
| | i ype of reaction. |